

260(5) : Basic Equations of the Structure of Matter

The starting equation is :

$$\underline{\nabla} \times \underline{v^a} = K \underline{v^a} - (1)$$

where $\underline{v^a}$ is a tetrad. This can be derived from
Cartan geometry. This equation implies that :

$$\underline{\nabla} \cdot \underline{v^a} = 0 - (2)$$

The ECE hypothesis is :

$$\underline{A^a} = A^{(0)} \underline{v^a} - (3)$$

so eq. (1) implies :

$$\underline{\nabla} \times \underline{A^a} = K \underline{A^a} - (4)$$

$$\underline{\nabla} \cdot \underline{A^a} = 0 - (5)$$

and

Eq. (1) implies :

$$(\nabla^2 + K^2) \underline{v^a} = 0 - (6)$$

so $\underline{v^a}$ has a large number of important solutions.
Using the minimal prescription and the Angie
hypothesis :

$$\underline{P} = e \underline{A} = \underline{k} \underline{K} - (7)$$

The quantum hypothesis is :

$$\underline{p} \phi = -i \hbar \nabla \phi \quad (8)$$

So

$$\underline{p} = eA^{(0)} \underline{\varphi} = \hbar k = -i \hbar \nabla \quad (9)$$

Now we:

$$eA^{(0)} = \hbar k \quad (10)$$

to find that:

$$\boxed{-i \nabla \phi = k \underline{\varphi} \phi} \quad (11)$$

For each a , this "quantization condition for the tetrad.

The kinetic energy is:

$$T = \frac{\underline{p}^2}{2m} = \frac{\hbar^2 k^2}{2m} \underline{\varphi}^2 = -\frac{\hbar^2 \nabla^2}{2m} \underline{\varphi}^2 \quad (12)$$

so

$$(\nabla^2 + k^2) \underline{\varphi} = 0 \quad (13)$$

and as in HFT 259:

$$(\nabla^2 + k^2) \phi = 0 \quad (14)$$

The Hamiltonian is defined by:

$$H = \nabla^2 + T \quad (15)$$

Here ∇ is the potential energy, so:

$$\boxed{k^2 = \frac{2m}{\hbar^2} (E - V)} \quad (16)$$

3) In quantum mechanics the wave function is normalized:

$$\int \psi^* \psi d\tau = 1. \quad (17)$$

so the solution of eq. (14) must be normalized, where

$$d\tau = r^2 \sin\theta d\theta d\phi. \quad (18)$$

The expectation value of the tetrad is:

$$\langle \underline{a} \rangle = \int \psi^* \underline{a} \psi d\tau. \quad (19)$$

The expectation value of any quantity is:

$$\langle \Omega \rangle = \int \psi^* \Omega \psi d\tau, \quad (20)$$

for example position:

$$\langle \underline{r} \rangle = \int \psi^* \underline{r} \psi d\tau. \quad (21)$$

This set of equations can now be applied to the general structure of the electron, proton and neutron.

In the special case of a Coulomb potential it is well known hydrogenic wave functions can be used. These are products of spherical harmonics and a radial function. In this case:

$$\nabla^2 = -\frac{e^2}{4\pi\epsilon_0 r}. \quad (22)$$

and for eqs. (16) and (21):

$$k^2 = \frac{2m}{\epsilon^2} \left(E + \frac{e^2}{4\pi \epsilon_0 r} \right) - (23)$$

It seems that the neutron has an empty internal structure and a finite magnetic dipole moment. The internal structure of the neutron will however be governed by a Deltram equation and Helmholtz equation of type:

$$(\nabla^2 + k^2)\phi = 0 - (24)$$

and its internal structure is given by the expectation value of position eq. (21), w.t. ϕ given by eq. (24). As we note 260(4), eq. (8), the simplest solution of eq. (24) in three dimension is:

$$\phi = j(r) Y(\theta, \phi) - (25)$$

$$\phi^* = j^*(r) Y^*(\theta, \phi) - (26)$$

so where $j(r)$ is a Bessel function and $Y(\theta, \phi)$ a spherical harmonic.

So the internal structure of the electron, proton and neutron will be given by eqs. (21), (25) and (26), w.t. $d\tau$ defined by eq. (18). These structures can be worked out by computer algebra.

The Bessel function $j(r)$ is the solution

3) of the Bessel differential equation:

$$\left(\frac{d^2}{dr^2} + \frac{2}{r} \frac{d}{dr} + \left(\kappa^2 - \frac{\ell(\ell+1)}{r^2} \right) \right) j(r) = 0 \quad -(27)$$

and the spherical harmonic is the solution of:

$$\frac{1}{\sin \theta} \frac{d \sin \theta}{d\theta} \frac{dy}{d\theta} + \frac{1}{\sin^2 \theta} \frac{d^2 y}{d\phi^2} = -\ell(\ell+1)y \quad -(28)$$

Unlike the hydrogen wavefunctions, it is a constant. The complete wavefunction must be normalized:

$$\int \psi^* \psi d\tau = 1. \quad -(29)$$

The latest data for the electron show that it has no electric dipole moment and is a perfect sphere (Imperial College experiment). This result refutes the standard model almost completely.

In this new ECE theory the dipole moment can be represented most simply by:

$$\underline{\mu} = e \underline{r} \quad -(30)$$

where e is the charge of the electron. So:

$$\langle \underline{\mu} \rangle = e \int \psi^* \underline{r} \psi d\tau \quad -(31)$$

6) Therefore the electron must be described by:

$$\int \psi^* \underline{S} \psi d\tau = 0 \quad - (32)$$

and the wave function ψ must be chosen to give this result.

On the other hand the electron has a magnetic dipole moment due to its intrinsic spin angular momentum, despite the fact that it appears to have a very tiny volume from the Imperial College experiment. The spin angular momentum operator is:

$$\hat{\underline{S}} \psi = \frac{\hbar \sigma}{2} \psi \quad - (33)$$

and the magnetic dipole moment operator is:

$$\hat{\underline{m}} = -g_e \frac{e}{2m} \hat{\underline{S}} \quad - (34)$$

where g_e is the g factor of the electron. So

$$\langle \underline{S} \rangle = \int \psi^* \underline{S} \psi d\tau = \frac{\hbar \sigma}{2} \quad - (35)$$

In the Z axis:

$$O_2 = \begin{bmatrix} 1 & 0 \\ 0 & -1 \end{bmatrix} \quad - (36)$$

Therefore wave function ψ must be chosen to give this result.

Finally the structure of the electron appears to be an almost infinitesimally small perfect sphere. To describe this result an orbital must be chosen to give a homogeneous result. The analogous result is the H atom is the ls orbital.

So the various properties of the electron are described by a choice of solutions ψ of the Helmholtz wave equation.

Finally, if we note the mass of the electron is described by the wave equation of ECE theory:

$$(\nabla^2 + \kappa_0^2) \frac{\psi}{m_e} = 0 \quad (37)$$

where

$$\kappa_0^2 = q^2 a^2 / (\omega_{\mu\nu}^2 - \Gamma_{\mu\nu}^2) \quad (38)$$

$$\rightarrow \left(\frac{mc}{\hbar} \right)^2$$

As in previous notes:

$$\frac{\omega^2}{c^2} = \kappa^2 + \kappa_0^2 \quad (39)$$

which is the Einstein energy equation.